## **Critical Review**

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# Rosa Baeza\* and Jorge Chirife Anthocyanin content and storage stability of spray/freeze drying microencapsulated anthocyanins from berries: a review

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Abstract: A comprehensive literature search for articles published on spray and freeze-dried anthocyanins from a large variety of berries was performed. Out of a total of twohundred and eight collected values, anthocyanin content in encapsulates had a 120-fold variation depending on the raw material and type of encapsulating agents. Highest observed anthocyanin concentration amounted to about 3500 mg/100 g powder. In most cases increasing the amount of encapsulant agents led to a noticeable reduction in the concentration of anthocyanins, this being attributable to a predominance of the dilution effect. Retention of encapsulated anthocyanins after storage at 25 °C (in darkness) for periods between 90 and 180 days were in the range of 80–67%, as long as the water activity  $(a_w)$  was 0.33 or less. Some predicted values of half-time  $(t_{1/2})$  from literature must be taken with precaution since in many cases they were derived from experimental measurements taken at storage times smaller than predicted half times. Anthocyanin degradation during storage occurred even below the glass transition temperature (T<sub>o</sub>) of the amorphous matrices.

**Keywords:** anthocyanins; berries; freeze-drying; microencapsulation; spray-drying; storage stability; water activity.

## **1** Introduction

Colorants for food applications obtained from natural sources are an attractive alternative to consumers [1, 2].

Anthocyanins form a group of pigments responsible for the red, purple, and blue colors and have potential as colorants. They are water-soluble pigments found in many fruits, flowers and leaves and the most abundant anthocyanins are pelargonidin, cyanidin, delphinidin, peonidin, petunidin and malvidin [3]. The interest in the use of natural colorants goes beyond their pigmentation capacity, since these compounds have bioactive properties that may be beneficial for consumers health [4].

It is well known that anthocyanins may undergo degradation during processing and storage; however, their stability may be increased through encapsulation. The encapsulated material is called core material, and the packaging material is called carrier, or encapsulating agent. Encapsulating agents act as protecting coat against factors such as light, humidity and oxygen, and, maltodextrin, Arabic gum, and modified starches are among most used carriers [5]. Several reviews about encapsulation techniques are available in the literature and the reader is referred to them [6–9]. Spray-drying is the most common technique used to encapsulate anthocyanins [9]. Freeze-drying is also an efficient method for anthocyanin encapsulation to produce a porous structure especially useful for temperature-sensitive bioactive products like anthocyanins [10].

One of the important conditions for an encapsulated anthocyanin to be used as a colorant is that it has a relatively high concentration in the powder resulting from spray/freeze-drying. For this, and other factors, it is necessary to minimize the addition of encapsulant to avoid a dilution effect. The amount of encapsulant is associated with the protection of anthocyanins and with maintaining desired flow properties of the powder product (absence of stickiness, caking or collapse) during dehydration and storage. For this reason, an elevated glass transition temperature ( $T_g$ ) is also required for encapsulants [11].

Anthocyanins are found in various small fruits such as grapes, blueberry, blackberry, strawberry, blackcurrant, and many other berries. In the past two decades a large number of articles on microencapsulation of berries have

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been published. It is the purpose of present review to compile and analyze a huge number of articles with appropriate data on, a) content of total anthocyanins in the dry powders, and b) the storage stability of anthocyanins at 25 °C. The stability of the anthocyanins refers to the observed retention at the end of the storage period and effect of type of encapsulating agents and water activity on length of storage stability is also discussed.

# 2 Anthocyanin content in powders

## 2.1 Data collection from the literature

A huge amount of literature data on anthocyanin encapsulation was initially reviewed, but only those papers that showed data on anthocyanin concentration and encapsulant/core ratio, were chosen. Table 1 summarizes literature data compiled on microencapsulation – via spray-drying or freeze-drying – of a wide variety of berries which were obtained from 50 eligible articles with appropriate data. The level of addition of encapsulant/s, and anthocyanin content in the final dry powder (mg anthocyanin/100 g powder) were collected. In the vast majority of cases the anthocyanin concentration is expressed as Cy3-glu equivalent, as usually done in berry fruits [12].

The concentration of anthocyanins in the encapsulates (obtained via spray drying or freeze-drying) depends on various factors; a) the anthocyanin concentration in the raw material, b) the efficiency (i.e., anthocyanin retention) of the encapsulation process and, c) the amount of encapsulant added. Type and amount of encapsulating agent serves to protect anthocyanins and also increase the glass transition temperature ( $T_g$ ) of encapsulated powder by preventing stickiness, caking and collapse, either during drying or storage [11]. However, the amount of encapsulant also leads to dilution of final anthocyanin concentration.

Two hundred and eight anthocyanin concentration values were obtained from Table 1 corresponding to different raw materials encapsulated under different conditions. It is not easy to say what anthocyanin concentration would be needed to be useful as a natural colorant, but it is reasonable to assume that low values will not be useful from a practical point of view. The compiled literature values of anthocyanin content ranged from about 30 to 3600 mg/100 g powder, depending on raw material, encapsulation method and type and amount of encapsulating agent/s. This range represents a 120-fold variation in anthocyanin content of encapsulated powders.

Figure 1 shows a Pareto chart for anthocyanin concentration in the encapsulates; this chart contains both bars and a line graph, where individual values are represented in descending order by bars, and the cumulative total is represented by the line. Although arbitrarily, anthocyanin concentration values can be ordered in ranges, e.g., low: up to 600 mg/100 g powder; intermediate: 600 to 1500 mg/100 g powder and high: more than 1500 mg/100 g powder. It can be seen (Figure 1) that anthocyanin values up to 600 mg/100 g powder represent 65% of literature values collected.

Several authors have reported anthocyanin concentration data according to whether they have used spray or freeze drving while all other variables (raw material, amount and nature of encapsulants) were kept identical. It is interesting to compare the values obtained by one or the other of these encapsulation methods. Michalska et al. [13] reported anthocyanin content after encapsulation of blackcurrant juice (12.4 °Brix) using maltodextrin, inulin and their mixtures as encapsulants. The mean value of anthocyanin concentration in nine samples encapsulated by freeze-drying was only 20% higher than those corresponding to spray drying. Murali et al. [14] also reported anthocyanin content in black carrot juice (6 °Bx) encapsulated by freeze-drying and spray-drying using identical encapsulants (maltodextrin, Arabic gum and tapioca starch) in the same proportions. The average of anthocyanins values in three samples encapsulated via freezedrying was 16.6% higher that corresponding to identical samples dried by spray. These results suggest that in some cases spray drying allows obtaining anthocyanin concentrations that are not much lower than those obtained in freeze-drying.

As shown in Table 1, the maximum total anthocyanin concentration values were about 3500 mg/100 g powder and corresponded to Aronia berry extract (ultrasound assisted) spray-dried with maltodextrin [15] and to açai juice spray-dried with maltodextrin [16].

### 2.2 Effect of addition of encapsulant agents

The glass transition temperature is mainly a function of water content, molecular weight and nature of the dry matter compounds. Berries are rich in low molecular weight sugars (fructose and glucose) and organic acids (citric, malic and tartaric acids). As noted by Das and Jaya [17], these compounds have low glass transition temperatures. Khalloufi et al. [18] reported T<sub>g</sub> values for various berries at "zero" (freeze-dried powdered berries

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 Table 1: Compilation of literature data on anthocyanin concentration after microencapsulation – via spray drying or freeze-drying – for a wide variety of berries and other anthocyanin-rich materials.

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Encapsulation method	Raw material	Encapsulant	% Encapsulating agent (or ratio extract)	Total anthocyanins (mg Cy3-glu/100 g powder)	Reference
Freeze-drying	Blueberry by-	MD DE10	30% in 8% concentrate	297.9	Yamashita et al. [46]
, .	product extract	MD DE20	30% in 8% concentrate	265.7	
Sprav drving	Blackcurrant iuice.	Maltodextrin DE20-40	30%	172.6	Michalska et al. [13]
	12.4 °Bx	Maltodextrin DE20-40	35%	139.1	
		Maltodextrin DE20-40	40%	99.0	
		MD DE20-40:Inulin (3:1)	30%	198.1	
		MD DE20-40:Inulin (3:1)	35%	165.8	
		MD DE20-40:Inulin (3:1)	40%	137.0	
		Inulin	30%	190.1	
		Inulin	35%	167.4	
		Inulin	40%	190.1	
Freeze-drving	Blackcurrant iuice.	Maltodextrin DE20-40	30%	211.0	Michalska et al. [13]
	12.4 °Bx	Maltodextrin DE20-40	35%	180.6	
		Maltodextrin DE20-40	40%	211.0	
		MD DE20-40:Inulin (3:1)	30%	224.7	
		MD DE20-40:Inulin (3:1)	35%	182.8	
		MD DE20-40:Inulin (3:1)	40%	152.3	
		Inulin	30%	233.3	
		Inulin	35%	198.3	
		Inulin	40%	155.6	
Freeze-drying	Elderberry pulp,	MD DE10	4%	1059	Baeza et al. [38]
, .	10.7 °Bx	Capsul™	4%	1200	
		Promitur™	4%	1251	
		k-Carrageenan	2%	1363	
Freeze-drying	Sour cherry pomace extract	Soy protein isolate	33%	92	Saponjac et al. [47]
Freeze-drying	Cherry juice, 18.7% °Bx	MD10 + A. gum (80:20)	20%	67.5	Sanchez et al. [48]
Freeze-drying	Maqui berry extract	MD DE15	10%	1185.7	Romero-González et al. [49]
		MD DE15	20%	662.4	
		MD DE15	30%	419.4	
		Inulin	10%	1021.4	
		Inulin	20%	835.4	
		Inulin	30%	446.2	
		Arabic gum	10%	481.3	
		Arabic gum	20%	309.3	
		Arabic gum	30%	466.0	
		MD DE15 + Inulin (1:1)	10%	1118.1	
		MD DE15 + Inulin (1:1)	20%	773.5	
		MD DE15 + Inulin (1:1)	30%	237.7	
		MD DE15 + A. Gum (1:1)	10%	530.8	
		MD DE15 + A. Gum (1:1)	20%	312.6	
		MD DE15 + A. Gum (1:1)	30%	367.4	
		MD DE15 + A.	10%	476.4	
		Gum + mum (1:1:1) MD DE15 + A. Gum + Inulin (1·1·1)	20%	341.3	
		MD DF15 + A	30%	565 1	
		Gum + Inulin (1:1:1)	2070	505.1	
Freeze-drying		MD:Soy protein isolate (2:1)	2.6-10.4%	330	Fredes et al. [50]

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### Table 1: (continued)

Encapsulation method	Raw material	Encapsulant	% Encapsulating agent (or ratio extract)	Total anthocyanins (mg Cy3-glu/100 g powder)	Reference
Spray drying	Maqui juice,	MD:Soy protein isolate	2.6-10.4	330	
	concentrated, 65 °Bx	(2:1)			
Freeze-drying	Blueberry pulp, 9 °Bx	MD DE10:A. Gum (80:20)	20%	74.3	Busso Casati et al. [51]
	Elderberry pulp, 11.8 °Bx	MD DE10:A. Gum (80:20)	20%	171.8	
	Maqui berry pulp, 14.8 °Bx	MD DE10:A. Gum (80:20)	20%	421.8	
	Blackcurrant pulp, 13.0 °Bx	MD DE10:A. Gum (80:20)	20%	254.6	
Freeze-drying	Jauboticaba by-	MD10	NR	273	Rodríguez et al. [52]
<i>,</i>	product extract	MD10:Guar Guam (0.95:0.05)	NR	265	
Spray drying	Chokeberry extract, 15 °Bx	MD DE18-20 + Arabic	1.2 g carrier/1 g choke- berry extract	1940.2	Pieczykolan and Kurek [53]
		MD DE18-20 + Inulin	1.2 g carrier/1 g choke-	2730.8	[]
			berry extract		
		MD DE18-20 + B-Glucan	1.2 g carrier/1 g choke- berry extract	3055.6	
		MD DE18-20 + Pectin	1.2 g carrier/1 g choke- berry extract	2853.8	
		MD18-20 + Guar Gum	1.2 g carrier/1 g choke- berry extract	2319.8	
Spray drying	Blackcurrant extract from waste	Maltodextrin DE12	40%	63.0	Archaina et al. [54]
Spray drying	Aronia berry extract	Maltodextrin 5.9	20%	3200	Vidović et al. [15]
(120/80 °C)	(ultrasound-	Maltodextrin 5.9	40%	2977	
	assisted)	Maltodextrin 5.9	60%	2764	
		Maltodextrin 13.1	20%	3313	
		Maltodextrin 13.1	40%	2828	
		Maltodextrin 13.1	60%	2785	
		Maltodextrin 19.7	20%	3511	
		Maltodextrin 19.7	40%	3119	
Convey day in a	Diask servete	Maltodextrin 19.7	60%	2747	Fraus and Vurgedal [6.6]
Spray drying	extract, 6 °Bx	Mattodextrin DE 28-31		480	Ersus and furgadet [44]
		Maltodextrin DE 20-23	to 20%	630	
Enroy drying	Eldorborny co. 2.0%			500	Cognoton at al [EE]
Spray urying	Blackcurrant, ss. 2.9%	MD DE12 MD DE12	20%	125	Gagneten et al. [55]
	Raspberry, ss.	MD DE12	20%	50	
Spray drying	Acai iuice	MD DE 10	6%	3436.9 <sup>b</sup>	Tonon et al. [16]
op.a) a.j3		MD DE 20	6%	3402.3 <sup>b</sup>	
		Arabic gum	6%	3416 <sup>b</sup>	
		Tapioca starch	6%	3247.2 <sup>b</sup>	
Spray drying	Blackberry pulp, 6.43% sugars	Maltodextrin	7%	666.0	Ferrari et al. [42]
		Arabic gum	7%	545.1	
		Maltodextrin:A. Gum (1:1)	7%	637.3	

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### Table 1: (continued)

Encapsulation method	Raw material	Encapsulant	% Encapsulating agent (or ratio extract)	Total anthocyanins (mg Cy3-glu/100 g powder)	Reference
Spray drying	Chokeberry juice,	Maltodextrin DE10	Juice (db):Carrier (2:3)	2002	Bednarska and Janiszewska-Turak [56]
	12 07	Maltodextrin DE15	ldem	1830	Jamszewska-Turak [50]
			Idem	2028	
		A.Gum.MD10 $(1.3)$	Idem	1070	
		A.Gum.MD10 (1.1)	Idem	1870	
		A.Gum-MD15 (3.1)	Idem	1007	
		A.Gum:MD15 $(1.5)$	Idem	1927	
		A.Guin: $MD15(1:1)$	1dem	1644	
		A. Guill:WD15 (5:1)	Idem	1400	
Courses double a		Arabic guili Dalaria darama lidara		1936	De France Markada
Spray drying	dues extract, 8.6 °Bx	Polyvinylpyrrolldone	ratio of 5:2	136	et al. [57]
Spray drying	Jucara fruit pulp	MD DE18:Arabic gum (1:1)	Pulp:Carrier (2:3)	151.7	Mazuco et al. [58]
Freeze-drying		MD DE18:Arabic gum (1:1)	Pulp:Carrier (2:1)	116.9	
		MD DE18:Arabic gum (1:1)	Pulp:Carrier (2:3)	150.8	
Spray drying	Blackberry extract	Arabic gum	10%	1280	Rigon and Zapata Noreña [59]
		Arabic gum	15%	1260	
		Polydextrose	10%	922	
		Polydextrose	15%	905	
Spray drying	Blackcurrant extract, 12 °Bx	Maltodextrin 11	to 35 °Bx	452.8	Bakowska-Barczaka and Kolodziejczy [60]
		Maltodextrin 18	to 35 °Bx	451.3	
		Maltodextrin 21	to 35 °Bx	430.4	
		Inulin	to 35 °Bx	398.9	
Spray drying	Bayberry juice, 11 °Bx	Maltodextrin DE10	1:1 (juice:MD at 11 °Bx)	56.4	Fang and Bhandari [29]
Spray drying	Mao fruit juice	Maltodextrin DE10-12	25%	334	Suravanichnirachorn et al. [61]
		Maltodextrin DE10-12	30%	271	
		Maltodextrin DE10-12	35%	263	
		Arabic gum	25%	347	
		Arabic gum	30%	286	
		Arabic gum	35%	234	
Spray drying	Blueberry juice	MD18/Whey Prot. Isolate = 0.4	10%	434	Darniadi et al. [62]
		MD18/Whey Prot. Isolate = 1.0	10%	484	
		MD18/Whey Prot. Isolate = 1.6	10%	513	
		MD18/Whey Prot. Isolate = 2.3	10%	529	
		MD18/Whey Prot. Isolate = 3.2	10%	538	
Foam-mat freeze-drying	Blueberry juice	MD18/Whey Prot. Isolate = 0.4	5%	711	Darniadi et al. [62]
		MD18/Whey Prot. Isolate = 1.0	5%	726	
		MD18/Whey Prot. Isolate = 1.6	5%	728	

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### Table 1: (continued)

Encapsulation method	Raw material	Encapsulant	% Encapsulating agent (or ratio extract)	Total anthocyanins (mg Cy3-glu/100 g powder)	Reference
		MD18/Whey Prot.	5%	785	
		Isolate = 2.3 MD18/Whey Prot.	5%	809	
		Isolate = 3.2			
Spray drying	Corozo fruit extracts (a) & (b)	Maltodextrin 18	Extract (a)/MD = 1:1	118	Osorio et al. [63]
		Maltodextrin 18	Extract (b)/MD = 1:1	131	
Spray drying	Maqui juice + extract	MD19:A. Gum (1:1)	10%	1938	Bastías-Montes et al. [64]
Spray drying	Andes berry 9 °Bx	MD20	Aq. extract:Carrier (1:1)	54.5	Villacrez et al. [65]
. , , , ,	•	Corn starch	Aq. extract:Carrier (1:1)	30.2	
		Yucca starch	Aq. extract:Carrier (1:1)	38.7	
		Capsul <sup>®</sup> TA	Aq. extract:Carrier (1:1)	46.0	
		Hi-CapTM 100	Aq. extract:Carrier (1:1)	59.4	
		MD DE 20/Yucca starch (1:1)	Aq. extract:Carrier (1:1)	58.1	
Spray drying	Blueberry conc. extract. 35% ss	Mesquite gum 17%	M. Gum 17%:Extract 35% (67:33)	1495	Jimenez-Aguilar et al. [28]
Spray drying	Bayberry syrup (crushed)	Maltodextrin DE10 (13%)	MD:Syrup (1:1)	80	Cheng et al. [66]
Freeze-drying		Maltodextrin DE10 (13%)	MD:Syrup (1.1)	121	
Spray drying	Black carrot extract	Starch, Soy protein, emulsifier	NR	3000	Mishra et al. [33]
		Starch, Soy protein, emulsifier	NR	2750	
Spray drying	Cranberry juice, 15% ss	MD DE11	15%	79.5	Zhang et al. [67]
		MD DE18	15%	74.9	
		Arabic gum	15%	86.4	
		A. Gum:MD DE11 (1:1)	15%	102.8	
Spray drying	Chokeberry pulp juice	HP-B-cyclodextrin	15%	307	Wilkowska et al. [25]
		Maltodextrin DE14-17	15%	427	
		A. Gum:MD DE14-17 (1:1)	15%	366	
Spray drying	Pomegranate juice, 16 °Bx	MD DE16	20.1%	74 <sup>a</sup>	Robert et al. [68]
		Soy protein isolate	7.5%	69 <sup>a</sup>	
Spray drying	Red raspberry puree	Arabic gum	10%	485 <sup>c</sup>	Syamaladevi et al. [69]
Spray drying	Barberry juice, 14 °Bx	MD12	15%	360	Sharifi et al. [70]
		MD DE12:A. Gum (1:1)	15%	380	
		MD DE12:A. Gum (3:1)	15%	360	
		Arabic gum	15%	350	
		MD DE12:A. Gum (1:3)	15%	355	
Freeze-drying	Blackberry pulp, 9 °Bx	Starch:Arabic gum (1:1)	1:1.78 (pulp solids/ encap.	126	Nogueira et al. [71]
Spray drying		Starch:Arabic gum (1:1)	Idem	86.4	
Freeze-drying	Blackberry juice, 5.4%	MD DE12	Juice:Matrix 80:20	162	Franceschinis et al. [72]
		MD DE12:Trehalose (1:1)	Juice:Matrix 80:20	112	
Spray drying	Blackberry juice, 5.4%	MD DE12	Juice:Matrix 60:40	70	
		MD DE12:Trehalose (1:1)	luice:Matrix 60:40	61	

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### Table 1: (continued)

Encapsulation method	Raw material	Encapsulant	% Encapsulating agent (or ratio extract)	Total anthocyanins (mg Cy3-glu/100 g powder)	Reference
Spray drying	Barberry juice, 16 °Bx	MD DE6	4%	1110	Emam-Djomeh et al. [73]
		MD DE6	6%	1060	
		MD DE6	8%	1070	
Spray drying	Mulberry juice	MD:A. Gum:MCC (21:9:0)	30%	468	Do and Nguyen [74]
		MD:A. Gum:MCC (21:8.5:0.5)	30%	341	
		MD:A. Gum:MCC (21:8:1)	30%	356	
		MD:A. Gum:MCC (21:7.5:1.5)	30%	330	
		MD:A. Gum:MCC (21:7:2)	30%	319	
Spray drying	Red grape marc extract, s.s. 2.97%	Maltodextrin DE18	Encapsulant:Extract 1:1	210	Moreno et al. [75]
	,	Maltodextrin DE18	Encapsulant:Extract 0.8:1	280	
		Maltodextrin DE18	Encapsulant:Extract 0.5:1	340	
		Maltodextrin DE18	Encapsulant:Extract 0.25:1	350	
		Maltodextrin DE18	Encapsulant:Extract 0.1:1	400	
		Whey protein isolate	Encapsulant:Extract 1:1	120	
		Whey protein isolate	Encapsulant:Extract 0.5:1	260	
		Whey protein isolate	Encapsulant:Extract 0.25:1	350	
		Whey protein isolate	Encapsulant:Extract 0.1:1	410	
		Pea protein isolate	Encapsulant:Extract 1:1	250	
		Pea protein isolate	Encapsulant:Extract 0.5:1	270	
		Pea protein isolate	Encapsulant:Extract 0.25:1	420	
		Pea protein isolate	Encapsulant:Extract 0.1:1	480	
Spray drying	Black carrots juice, 6 °Bx	Maltodextrin DE20	to obtain 20 °Bx in mixture	1461.2	Murali et al. [14]
		Tapioca starch	to obtain 20 °Bx in mixture	1085.8	
		Arabic gum	to obtain 20 °Bx in mixture	1381.5	
Freeze-drying		Maltodextrin DE20	to obtain 20 °Bx in mixture	1650	
		Tapioca starch	to obtain 20 °Bx in mixture	1353.5	
		Arabic gum	to obtain 20 °Bx in mixture	1577.3	
Spray drying	Blackcurrant extract 12 °Brix	MD10	to obtain 35 °Bx in mixture	452.8	Bakowska-Barczaka and Kolodziejczyk [60]
		MD18	to obtain 35 °Bx in mixture	451.3	
		MD21	to obtain 35 °Bx in mixture	430.4	
		Inulin	to obtain 35 °Bx in mixture	398.9	

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Table 1: (continued)

Encapsulation method	Raw material	Encapsulant	% Encapsulating agent (or ratio extract)	Total anthocyanins (mg Cy3-glu/100 g powder)	Reference
Spray drying	Garcinia indica	Maltodextrin DE6	5%	325.3	Nayak and Rastogi [7]
	Choisy fruit extract,	Maltodextrin DE19	5%	365	
	hhh6 °Bx	Maltodextrin DE21	5%	430.3	
		Maltodextrin DE33	5%	399.0	
Spray drying	Bordo grape skin extract	Gum Arabic:Polydextrose (1:1)	10%	1718 <sup>a</sup>	Kuck and Noreña [76]
		Arabic gum	10%	1841 <sup>a</sup>	
		Partially hydrolyzed gum guar (PHGG)	10%	2084 <sup>a</sup>	
		PHGG:Polydextrose	10%	2105 <sup>a</sup>	
Freeze-drying		Gum Arabic:Polydextrose	10%	1838 <sup>a</sup>	
		Arabic gum	10%	1727 <sup>a</sup>	
		PHGG	10%	1,2, 1825 <sup>a</sup>	
		PHGG·Arabic gum (1·1)	10%	1707 <sup>a</sup>	
Spray drying	Maqui berry extract,	SiO <sub>2</sub>	Soluble solids:SiO <sub>2</sub>	1514 (average)	Garrido Makinistian et al.
	10.1 °Bx		(1:0.5)		[77]
Spray drying	Blackberry extract, 13.4 °Bx	Maltodextrin DE10	22.5 g Encap:7.5 g solids juice	87.2	Diaz et al. [78]
		Arabic gum	Idem	106.9	
		Whey protein concentrate	Idem	34.1	
Spray drying	Pomegranate juice, 16.5 °Bx	Modified starch Capsul™	16.50%	75.8	De Araujo Santiago et al. [79]
		Maltodextrin DE6	16.5%	56.4	
		Gum Arabic	16.5%	94.8	
		Capsul + Maltodextrin	16.5%	70.4	
		Arabic gum + Capsul	16.5%	111.5	
		Arabic gum + maltodextrin	16.5%	84.5	
Spray drying	Grape juice, 14 °Bx	Soy protein-maltodextrin	1.25 g encap./g sol. Solids	717 <sup>d</sup>	Moser et al. [80]
		Soy protein-Maltodextrin	1.0 g encap/g soluble solids	677	
		Whey protein-	0.75 g encaps./g sol.	815	
		Maltodextrin	Solids		
		Whey protein	0.85 g encap./g sol.	776	
		maltodextrin	Solids		
Spray drying	Blueberry waste	Sodium alginate	2%	1954	Waterhouse et al. [26]
	extracts	Inulin	2%	1821	

<sup>a</sup>Total anthocyanins expressed as mg malvidin-3-glucoside.<sup>b</sup>Total anthocyanins expressed as mg cyanidin-3-rutinoside/100 g powder.<sup>c</sup>Total anthocyanins expressed as mg malvidin-3,5-diglucoside/100 g powder.<sup>d</sup>Total anthocyanins expressed as mg malvidin-3,5-diglucoside/100 g powder.

dehydrated to almost "zero" moisture by keeping them in desiccators under vacuum and over desiccant at least for a week) or very low moisture values (3–4%). Reported  $T_g$  values were relatively low since for all berries having a moisture content of 0% the they ranged between 26 and 45 °C, while for berries having around 3% moisture  $T_g$ 

values did not exceed 20 °C. Similar low values were also reported by Staniszewska et al. [19], Syamaladevi et al. [20] and Moraga et al. [21] for low moisture cranberry, raspberry and strawberry. This justifies the need of addition of high  $T_g$  encapsulants (i.e., maltodextrins, Arabic gum and the like). Saavedra-Leos et al. [22] determined

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Anthocyanin concentration (mg/ 100 g dry basis)

the glass transition temperature of a large number of ternary and binary mixtures of glucose, fructose and sucrose having different mass fractions mixtures resembling berries and obtained  $T_g$  values relatively close to those showed here.

Twenty eligible articles (from Table 1) had appropriate data to calculate the relation *r* defined as: g encapsulate/g of soluble solids in the juice (or extract). The values of *r* varied widely ranging from <1 to 22; the majority (69%) of them were in the range <1 to 4 and 17% in the range 8 to 22. It is well known in literature [23] that compatible blends of amorphous solids show a single  $T_g$  that is intermediate to the  $T_g$  values of the solids and is related to the mixture composition. When the additive raises the  $T_g$  it is to have an anti-plasticizing effect, and this the case when a biopolymer (i.e., maltodextrin) is used for encapsulation of a berry product. The relationship of  $T_{gmix}$  to the mixture composition is known as Gordon and Taylor equation (Eq. (1)),

$$T_{gmix} = \left[\frac{w_1 T_{g1} + K w_2 T_{g2}}{w_1 + K w_2}\right]$$
(1)

where  $w_1$  and  $w_2$  are mass fractions,  $T_{g1}$  and  $T_{g2}$  are individual  $T_g$  of components, and K is a constant. In some cases, Eq. (1) can be written as the Fox equation [24],

$$1/T_{gmix} = (w_1/T_{g1}) + (w_2/T_{g2})$$
 (2)

which is a very crude approximation to predict  $T_{gmix}$  only knowing the weight fraction and  $T_g$  of each component. Nevertheless Eq. (2) is useful to illustrate that increasing the mass fraction of encapsulant will increase the  $T_g$  of material to be encapsulated.

In terms of mass fraction, the mass fraction of encapsulating agents (i.e., g encapsulate/g soluble solids) can be calculated from the *r* values shown above, the most common

**Figure 1:** Pareto chart for frequency of anthocyanin concentrations in powders.

values of encapsulant mass fraction ranged from 0.29 to 0.80, but some values were as high as 0.90 or 0.956.

It is to note that a high mass fraction of encapsulant would not be compatible with the use as a potential natural colorant.

Figures 2 and 3 were constructed using data from Table 1 and show the effect of the amount of encapsulating agent (maltodextrin, Arabic gum or inulin) on anthocyanin content after encapsulation (spray drying or freeze drying) of different raw materials, namely, Aronia berry extract, maqui berry extract, mao fruit juice, red grape marc extract and blackcurrant juice. As observed, increasing encapsulant addition decreases anthocyanin concentration in the encapsulated material. Freeze-dried maqui berry (Figure 2a), shows the largest relative decrease in total anthocyanin content with increasing encapsulant addition. This can be attributed to the fact that freeze-drying "per se" is a beneficial process for temperature-sensitive products, so that the increase in addition of solutes favors dilution rather than protection during lyophilization. Thus, the least amount of encapsulant compatible with the desired anthocyanin protection and flow properties (absence of stickiness, caking or collapse) should be used to maximize anthocyanin concentration in the encapsulating powder.

# 3 Storage stability of microencapsulated anthocyanins

# 3.1 Compilation of literature data on storage stability

Table 2 summarizes literature data compiled on measured storage stability at 25 °C of encapsulated anthocyanins. The

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Figure 2: Effect of encapsulant addition on anthocyanin content of powders.

**a.** Aronia berry extract spray-drying encapsulated in maltodextrin (from data of Vidovic et al. [15]) – The standard deviation bars were 0.3–2% of the data shown in the figure. **b.** Maqui berry extract freeze-drying encapsulated in maltodextrin (from data of Romero-González et al. [49]) – The error bars were 0.4–4.4% of the data shown in the figure. **c.** Blackcurrant juice spray-drying encapsulated in maltodextrin (from data of Michalska et al. [13]) – The error bars were 1.6–4.2% of the data shown in the figure.

sources of anthocyanins comprised a large variety of berries which were obtained from eligible articles from literature reporting appropriate storage stability data. The anthocyanin stability is indicated by the calculated retention  $C_t/C_0 \times 100$ , where  $C_0$  is the initial anthocyanin concentration and  $C_t$  is the concentration at the end of studied storage period. Type of encapsulating agents, storage temperature and water activity ( $a_w$ ) or moisture content, are also indicated. Only works which stored the samples at least for 90 days were chosen for this survey.

Several observations can be made about the data shown in Table 2. Storage times studied vary widely from author to author, ranging from 90 to 360 days. Tonon et al. [16] reported that after 120 days at 25 °C and  $a_w = 0.328$ , the retention of acai anthocyanin encapsulated in maltodextrin DE10, maltodextrin DE20 or Arabic gum was very similar, namely 87, 80 and 83.5%, respectively. Wilkowska et al. [25] reported the storage stability of chokeberry anthocyanin encapsulated in maltodextrin, maltodextrin + Arabic gum or HP- $\beta$ -cyclodextrin at 25 °C during 360 days at 3% moisture content. Observed retentions of anthocyanin were more or less similar, namely 72.6, 70.6 and 78.8%, respectively.

Nayak and Rastogi [7] encapsulated *Garcinia indica* Choisy in five different encapsulants with moisture content from 4.4 to 5.8% at 25 °C. Encapsulants were, Maltodextrin DE6, DE19, DE21, DE33 and maltodextrin DE21 + gum acacia. After 98 days they observed anthocyanin retentions of 70, 72.4, 72.9, 72.7 and 80%, respectively indicating that



Figure 3: Effect of encapsulant addition on anthocyanin content of powders.

**a**. Mao fruit juice spray-drying encapsulated in Arabic gum (from data of Suravanichnirachorn et al. [61]) – The standard deviation bars were 7.0–9.0% of the data shown in the figure.

**b**. Maqui berry extract freeze-dried encapsulated in inulin (from data of Romero-González et al. [49]) – The error bars were 1.0–4.5% of the data shown in the figure.

the type of encapsulant did not influence the stability too much. The sample of maltodextrin DE21 + gum acacia which had the highest retention (80%) also had the lowest moisture content (4.4%) and this may have contributed to a somewhat higher retention than the other samples having higher moisture contents (5.0 and 5.8%). Waterhouse et al. [26] stored blueberry waste extract encapsulated with sodium alginate or inulin at 20 °C and  $a_w = 0.25-0.31$  for 90 days and calculated anthocyanin retentions were 94.5 and 89%, respectively. Pavón-García et al. [27] measured anthocyanin content during storage at 20 and 30 °C of muitle (Justicia spicigera) encapsulated in mixtures of mesquite gum, maltodextrin DE10 and Arabic gum having aw between 0.555 and 0.592. Little difference was observed in anthocyanin content after 120 days of storage and the calculated retention values were between 84.7 and 89.2% again suggesting that the type of encapsulant also played a minor role in observed storage stability. Jimenez-Aguilar et al. [28] reported that encapsulated (mesquite gum) blueberry extract stored at 4 and 25 °C had a higher stability at the lower temperature. Fang and Bhandari [29] studied the storage stability at 5 and 25 °C of encapsulated (maltodextrin DE10) of various water activities. After 3 months storage at  $a_w = 0.33$  calculated anthocyanin retention at 5 °C was 90%,

while at 25 °C it was 86%. After 6 months at 5 °C, retention was 86% while at 25 °C it was 81.7%. Mahdavi et al. [30] reported that encapsulated barberry extract using spraydrying and three different carriers: a combination of Arabic gum + maltodextrin, maltodextrin + gelatin and maltodextrin. The spray-dried encapsulated samples were stored at several temperatures for 90 days and they reported that the half-life (time for 50% retention) was significantly different between 4 and 25 °C.

## 3.2 Effect of exposure to light

It is known that native anthocyanins had poor stability to light and when exposed to natural light the retention of anthocyanins decrease rapidly [31]. Even with encapsulated anthocyanins, exposure to light during storage must be avoided since several authors reported that encapsulated anthocyanins stored in darkness (or amber-colored vials) were more stable than exposed to light. Wilkowska et al. [25] reported that black chokeberry encapsulated in maltodextrin and other encapsulants, were more stable stored in darkness than exposed to light. Patel et al. [32] encapsulated black carrots juice with starch, soy protein, whey protein and emulsifier, and found that samples were more stable stored

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### Table 2: Retention (%) of total microencapsulated anthocyanins observed after storage at 25 °C.

Anthocyanin source	Encapsulant/s and encapsulation method	Storage temperature, °C	Water activity (a <sub>w</sub> ) or moisture content (%)	Total anthocyanin retention (%) at the end of storage period (indicated)	Reference
Açai ( <i>Euterpe oleracea</i> Mart.)	Maltodextrin DE10- (spray drving)	25	a <sub>w</sub> = 0.328	87% at 120 days	Tonon et al. [16]
,	,	25	a <sub>w</sub> = 0.529	81.9% at 120 days	Tonon et al. [16]
	Gum arabic (spray-drying)	25	$a_w = 0.328$	83.5% at 120 days	Tonon et al. [16]
	Gum arabic (spray-drying)	25	$a_w = 0.529$	75.8% at 120 days	Tonon et al. [16]
	Maltodextrin DE20	25	$a_{w} = 0.328$	80% at 120 days	Tonon et al. [16]
	Maltodextrin DE20	25	a <sub>w</sub> = 0.529	74.0% at 120 days	Tonon et al. [16]
Chokeberry (Aronia	Maltodextrin (spray-drying)	25	3.2%	72.6% at 360 days	Wilkowska et al. [25]
melanocarpa) juice	Maltodextrin + Arabic gum (spray-drying)	25	3.5%	70.6% at 360 days	Wilkowska et al. [25]
	HP-β-cyclodextrin (spray- drving)	25	2.9%	78.8% at 360 days	Wilkowska et al. [25]
Bayberry juice	Maltodextrin DE10 (spray drving)	25	a <sub>w</sub> = 0.33	81.6% at 180 days	Fang and Bhandari [29]
	Maltodextrin DE10 (spray drving)	25	a <sub>w</sub> = 0.22	86.7% at 180 days	Fang and Bhandari
Blackcurrant extract	Maltodextrin DE11 (spray-	25	3.6%	79.8% at 360 days	Bakowska-Barczaka and Kolodzieiczyk [59]
	Maltodextrin DE18 (spray- drying)	25	3.9%	63.9% at 360 days	Bakowska-Barczaka and Kolodziejczyk [59]
	Maltodextrin DE 21 (spray- drying)	25	3.1%	70.7% at 360 days	Bakowska-Barczaka and Kolodziejczyk [59]
	Inulin (spray-drying)	25	3.8%	82.7% at 360 days	Bakowska-Barczaka and Kolodziejczyk [59]
Blueberry waste	Sodium alginate (spray-drying)	20	$a_{w} = 0.25$	94.5% at 90 days	Waterhouse et al. [26]
extract	Inulin (spray-drying)	20	$a_{w} = 0.31$	89% at 90 days	Waterhouse et al. [26]
Garcinia indica Choisy	Maltodextrin DE6 (spray drying)	25	5.8%	69.8% at 98 days	Nayak and Rastogi [7]
	Maltodextrin DE19 (spray drying)	25	5.7%	72.4% at 98 days	Nayak and Rastogi [7]
	Maltodextrin DE 21 (spray drying)	25	5.0%	72.9% at 98 days	Nayak and Rastogi [7]
	Maltodextrin DE33 (spray drying)	25	5.6%	72.7% at 98 days	Nayak and Rastogi [7]
	Maltodextrin DE21 + gum acacia and TCP (spray drying)	25	4.4%	80% at 98 days	Nayak and Rastogi [7]
<i>Hibiscus sabdariffa</i> extract	Maltodextrin DE11-15 + Arabic gum (spray drying)	25	as it comes out of dryer	65.7% at 104 days	Idham et al. [81]
Pomegranate juice	Arabic gum/Capsul (spray drying)	25	as it comes out of dryer	60% at 120 days	De Araujo Santiago et al. [79]
Muitle (Justicia spic- igera) aqueous extract	Mesquite gum/Maltodextrin DE10 (1:1), spray-drying	20	a <sub>w</sub> = 0.581	89.2% at 120 days	Pavón-García et al. [27]
	Maltodextrin DE10/Arabic gum (1:1), spray drying	20	a <sub>w</sub> = 0.555	84.7% at 120 days	
Blueberry juice	Maltodextrin DE5-7, spray drying	22	a <sub>w</sub> = 0.328	41% at 90 days	Turan et al. [82]
	Maltodextrin DE 5–7 + Arabic gum, spray drying	22	a <sub>w</sub> = 0.328	76% at 90 days	
Bordo grape wine- making pomace extract	Maltodextrin DE9-12, spray drying, 10% <sup>a</sup>	25	a <sub>w</sub> = 0.328	73.3 at 120 days	Souza et al. [45]
	Maltodextrin DE9-12, spray drying, 20% <sup>a</sup>	25	a <sub>w</sub> = 0.328	81.2% at 120 days	
	Maltodextrin DE9-12, spray drying, 30%ª	25	a <sub>w</sub> = 0.328	81.4% at 120 days	

 $^{\rm a}\textsc{Average}$  of 3 inlet air temperatures between 130 and 170 °C.

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in amber-colored vials tan transparent ones. Jimenez-Aguilar et al. [28] also reported that spray-dried blueberry extract encapsulated with mesquite gum and stored in presence of light increased the rate of anthocyanins degradation. Patel et al. [32] studied the degradation kinetics of encapsulated anthocyanin during storage in transparent as well as amber-colored vials at 37 °C. They found that the rate of degradation of anthocyanin was significantly higher in the microcapsules stored in transparent vials compared to those stored in amber-colored ones. Mishra et al. [33] encapsulated (starch/soy protein/whey protein) black carrot anthocyanins powder in amber colored bottles and transparent bottles and placed at 25 °C. After 63 days anthocyanin retention was 50% in amber colored bottles, but decreased to 35% in transparent bottles.

## 3.3 Water activity effects

As noted by several authors storage equilibrium relative humidity plays an important role on anthocyanin degradation of powders. The effect of water activity of powders can be explained by the fact that the degradation reaction of anthocyanins is controlled by the mobility of the reactants [16, 34]. The higher the water content (or water activity), the higher is the molecular mobility, which facilitates reactions of degradation. Selim et al. [35]. evaluated degradation kinetics of Roselle encapsulated anthocyanin under  $a_w$  of 0.43, 0.54, 0.64 and 0.75 and observed that the lowest degradation rate was observed at  $a_w = 0.43$  and increased when  $a_w$  was increased. Lavelli et al. [36] determined the kinetics of degradation of individual anthocyanin compounds in encapsulated extract of grape stored at 30 °C and different water activities and calculated corresponding half-times  $(t_{1/2})$ . Four selected anthocyanins from their data were plotted in Figure 4 as a function of a<sub>w</sub>. It can be seen that increasing a<sub>w</sub> from 0.22 to 0.75, decreased half-times dramatically (i.e., less stability). Figueiredo et al. [37] reported that storage stability of camu-camu anthocyanins encapsulated in maltodextrin, inulin or oligosaccharide was drastically reduced when water activity was increased from 0.22 to 0.75. Figure 5 is a plot of their half-times (time to decrease anthocyanin to 50%) showing that half-times decrease (or stability is increased) when a<sub>w</sub> is increased.

Fang and Bhandari [29] indicated that best condition to minimize degradation of encapsulated bayberry anthocyanin was an  $a_w$  below 0.33. Baeza et al. [38] found that decreasing  $a_w$  from 0.43 to 0.12–0.20 decreased the rate of anthocyanin degradation in freeze-drying encapsulated elderberry pulp.



**Figure 4:** Effect of water activity on half-time (days) for degradation of individual anthocyanin compounds in encapsulated (maltodextrin DE12) grape skin extract. (From data of Lavelli et al. [36]) – Mv-glc: malvidin 3-O glucoside; Cy-glc: cyanidin 3-O-glucoside; Pn-glc: peo-nidin 3-O-glucoside; Dp-glc: delphinidin 3-O-glucoside – Standard deviation bars for first-order rate constants (from which half-times were derived) were 1.2–4.0% of their values, with the exception of two data for which the standard deviation reached 10%.



**Figure 5:** Effect of water activity on half-life of anthocyanin degradation in encapsulated camu-camu extract (from data reported by Figueiredo et al. [37]) – MD: maltodextrin; Oligos: oligosaccharide – Authors did not provide error bars for their half-times.

Gradinarua et al. [39] reported kinetic studies of anthocyanins stability from *Hibiscus sabdariffa L*. (roselle) co-lyophilized with an amorphous polysaccharide (pullulan) stored in different relative humidity environments (water activities 0.33, 0.53, 0.75 and 0.84) at 40 °C. The rate constants for degradation were obtained from first-order reaction kinetic plots and the degradation rate constants increased with the water activity, particularly above 0.53.

## 3.4 Effect of glass transition (T<sub>g</sub>)

The glass transition theory has been also used to explain the effect of moisture on storage stability. According to theory, when a material is in the glassy state, the high viscosity of the matrix does not allow the occurrence of diffusion-controlled reactions [40]. However, some authors have demonstrated that some diffusion-controlled reactions, may occur, even at the glassy state [41]. Tonon et al. [16] reported that anthocyanin loss did occur in the glassy state of encapsulated acai. Ferrari et al. [42] reported that anthocyanin loss did occur in the glassy state of encapsulated acai or blackberry. Gradinarua et al. [39] measured the degradation kinetics of anthocyanins encapsulated by freeze-drying in pullulan and did not find any dependence on the molecular mobility of the system, as it relates to the glass-rubber transition. As they noted anthocvanin degradation occurred, even at sub-Tg temperatures of the amorphous matrices, whereas no changes in the rate constants were observed in the vicinity of the glass transition.

This contradicts the theory of Slade and Levine [40], according to which the storage at a temperature inferior to the glass transition temperature  $(T_g)$  assures product stability. The diffusion of oxygen may cause pigment degradation in the stored powders. According to Schebor et al. [43], factors like aging of the glassy material, rotational mobility and diffusion throughout pre-existing holes (pores), due to defects and porosity in the structure can explain the occurrence of chemical reactions in foods, even in the glassy state.

The occurrence of anthocyanin degradation at temperatures below the  $T_g$  points to the possibility of some sort of reactant mobility in the glassy state and indicates the insufficiency of using  $T_g$  as an absolute threshold of stability.

## 3.5 Kinetic analysis of anthocyanins during storage

Several authors had analyzed the stability of total anthocyanins in different systems using a kinetic approach, and degradation of encapsulated anthocyanins has been calculated according to the following equations,

$$\ln\left(C_{t}/C_{0}\right) = -k.t \tag{3}$$

where k is the rate constant,  $C_0$  is the initial anthocyanin content and  $C_t$  is the anthocyanin content after storage time t. The degradation constant (k) for the anthocyanin content was determined from slope of a plot of the natural log of anthocyanins retention ( $C_t$ /  $C_0$ ) versus time. The half-time ( $t_{1/2}$ ) (at a specific temperature) is determined as per the following equation:

$$t_{1/2} = \ln 2/k$$
 (4)

Knowledge of half-time is a simple way to compare the kinetics of anthocyanin degradation in different systems.

Time required for bioactive compounds to degrade to 50% of their original values (half-life) is an easy way to compare the storage stability of encapsulated anthocyanins and has been utilized by several workers. Table 3 shows a literature compilation of time required for encapsulated anthocyanins to degrade to 50% of their original value at 25 °C and specified water activity (or moisture content). Tonon et al. [16] found that anthocyanin degradation during storage of encapsulated acai exhibited two first-order kinetics with two values of k and  $t_{1/2}$ . Since halflife is determined as the time at which the anthocyanin content was reduced by 50% with respect to the zero-time, uncertainty appears in this case. It is to be noted that when the storage period studied cover a period of time much shorter than the predicted mean time, the risk of a failed extrapolation is important.

Ersus and Yurgadel [44] determined that the half-time at 25 °C of black carrot juice encapsulated with maltodextrin DE29 or Glucodry were identical (400 days) but when maltodextrin DE10 was used as encapsulant, the half-time was slightly reduced (360 days). The half-time for Garcinia indica Choisy encapsulated with maltodextrin of different dextrose equivalent, DE6, DE19, DE21 and DE33 [7] were 420, 450, 450 and 420 days respectively indicating that the encapsulant used did not change half-time too much. Similar behavior was reported by Madhavi et al. [30] for half-life times of barberry juice extract encapsulated in maltodextrin DE19, maltodextrin DE19 + Arabic gum, and maltodextrin DE19 + Gelatin. Souza et al. [45] indicated that increasing the added amount of maltodextrin DE10 increased the halftimes for encapsulated anthocyanins extracted from red grape residues extract. Data obtained by Pavón-García et al. [27] indicated that half-life of anthocyanins in a mixture of mesquite gum/maltodextrin was significantly higher than for a mixture of Arabic gum/maltodextrin.

## **4** Conclusions

Amount of encapsulants added, and content of total anthocyanins in the encapsulated dry powders, were analyzed in relation to potential of powders as natural food dyes. Over a total of two-hundred and eight values surveyed, anthocyanin encapsulates exhibited a 120-fold variation in their anthocyanin content, depending on the raw material and type and amount of encapsulating agents. The highest observed anthocyanin concentration amounted to about 3500 mg/100 g powder, corresponding to Aronia berry extract (ultrasound-assisted) and acai juice both spray-dried with maltodextrin.

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Anthocyanins source	Encapsulant/s	a <sub>w</sub>	Temp. (°C)	t <sub>1/2</sub> , days	Reference
Blackberry	MD DE20	0.259	25	374	Ferrari et al. [42]
	Arabic gum	0.314	25	297	
	MD DE20 + GA	0.278	25	348	
Black carrot juice	Maltodextrin DE29	As it comes out from dryer	25	400	Ersus and Yurdagel [44]
	Glucodry DE 20-23	As it comes out from dryer	25	400	
Roselle ( <i>Hibiscus sabdariffa</i> )	Maltodextrin	As it comes out from dryer	25	159	Idham et al. [81]
	Maltodextrin + A. gum	As it comes out from dryer	25	162	
	Arabic gum	As it comes out from dryer	25	141	
	Soluble starch	As it comes out from dryer	25	120	
Barberry extract	Maltodextrin DE19 + A. gum	As it comes out from drver	25	150	Madhavi et al. [30]
	Maltodextrin DE19 + Gelatin	As it comes out from drver	25	141	
	Maltodextrin DE 19	As it comes out from drver	25	141	
Bordo grape winemaking pomace extract	Maltodextrin DE10 (10%), spray drying <sup>a</sup>	a <sub>w</sub> = 0.328	25	271	Souza et al. [45]
	Maltodextrin DE10 (20%), spray drving <sup>a</sup>	a <sub>w</sub> = 0.328	25	403.2	
	Maltodextrin DE10 (30%), spray drying <sup>a</sup>	a <sub>w</sub> = 0.328	25	380	
Blueberry juice	Maltodextrin DE 4–7, spray drying	a <sub>w</sub> = 0.328	22	69.3	Turan et al. [82]
	Maltodextrin DE 4–7, spray drying	a <sub>w</sub> = 0.328	22	231	

Table 3: Literature predicted half-life for anthocyanin degradation in several encapsulates at near 25 °C and  $a_w \leq .$ 

<sup>a</sup>Average of 3 inlet air temperatures between 130 and 170 °C.

In most cases increasing the amount of encapsulant/s led to a noticeable reduction in the concentration of anthocyanins, this being attributable to a predominance of the dilution effect. Consequently, the least amount of encapsulant compatible with the desired anthocyanin protection and flow properties of powders (absence of stickiness, caking or collapse) should be used to maximize anthocyanin concentration since high mass fractions of encapsulants would not be compatible with the use of encapsulated anthocyanins as a colorant.

Several identical samples (raw material, type and quantity of encapsulant) encapsulated by spray drying or by freezedrying showed a difference of only 17–20% (in favor of freezedrying) in the amount of anthocyanins in the final powder.

After storage times between 90 and 180 days at 25 °C anthocyanin retentions were in the range 87–60% (in darkness) as long as the water activity was 0.33 or lower. In some systems the type of encapsulant played some role in the storage stability, while in others did not. It is noteworthy that some literature predicted values of half-time ( $t_{1/2}$ ) may be taken with precaution, since they were derived

from experimental measurements taken at storage times smaller than the predicted half times. Anthocyanin degradation occurred even at sub- $T_g$  temperatures of the amorphous matrices, which confirms the insufficiency of using  $T_g$  as an absolute threshold of stability.

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**Informed consent:** All the co-authors are fully aware and agree.r

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